

1 **Variable respiration rates from incubated permafrost soil extracts in the**
2 **Kolyma River lowlands region of Northeast Siberia**

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1 **Variable respiration rates from incubated permafrost soil extracts in the**
2 **Kolyma River lowlands region of Northeast Siberia**

3 Thawing permafrost soils supply dissolved organic carbon (DOC) to aquatic systems;
4 however, the magnitude, variability, and fate of this DOC is not well constrained. The
5 objective of this study was to examine respiration potentials from soil DOC derived from
6 seasonally-thawed and near-surface (<1.5 m) permafrost soils collected from five different
7 locations in the Kolyma River Basin, NE Russia. We measured soil bulk organic carbon
8 (OC) content, water-soluble macronutrients (DOC, NH₄, PO₄), and the heterotrophic
9 respiration potentials of the extract DOC in five-day laboratory incubations. DOC
10 concentrations in our soil extracts ranged from 2.8-27.9 mg L⁻¹ (mean 13.5 ± 2.5 mg L⁻¹, n
11 = 14). Mean carbon respiration was 0.13±0.08 mg C (-0.03-0.47 mg C, n=16) and 8.86-
12 31.35% total DOC (8.7-31.4%, n=14). While DOC concentration in the extracts was a
13 function of bulk soil OC concentration, we did not find a relationship between respiration
14 rates and soil OC or DOC concentrations. Respiration was highest in top active layer soils
15 (10-15 cm depth) but varied widely among sites. Respiration potentials were lowest at the
16 bottom of the seasonally-thawed active layer (30-50 cm depth), and C respiration of icy,
17 organic-rich Pleistocene-aged permafrost (yedoma) extracts varied across geographic
18 locations (0.04-0.47 mg C respired, 8.7-31.4% total DOC). Despite the small sample size,
19 our study indicates that near-surface soils and permafrost in the Kolyma River Basin are
20 spatially variable in terms of both soil OC content and soil extract respiration rates and that
21 OC contents do not predict C respiration rates. While a larger sample size would be useful
22 to confirm these results at broader geographic scales, these initial results suggest that soil
23 OC heterogeneity should be taken into account in efforts to determine the fate of soil OC
24 released from permafrost-dominated terrestrial ecosystems to aquatic ecosystems following
25 permafrost thaw.

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27 Keywords: arctic, carbon, permafrost, respiration, Russia, yedoma

1 **Introduction**

2 Perennially frozen ground (permafrost) contains a vulnerable carbon (C) pool susceptible
3 to warming and thaw (Zimov et al. 2006a; Schuur et al. 2008; Schuur et al. 2015). Permafrost
4 covers 22% of the Northern Hemisphere (Brown et al. 1998) and contains an estimated 1140-
5 1580 Pg of organic carbon (OC)— approximately half of the world’s below ground OC
6 (Hugelius et al. 2014; Schuur et al. 2015). Temperatures in the Arctic have increased an average
7 of 0.6°C per decade over the last 30 years, which is twice as fast as the global average (IPCC
8 2013). This climate warming triggers the release of permafrost OC via permafrost thaw and
9 erosion, exporting large amounts of terrestrial C to aquatic environments (Striegl et al. 2005;
10 Frey and McClelland 2009; Vonk et al. 2013; Cory et al. 2014; Larouche et al. 2015) and making
11 previously frozen OC from a range of soil depths available for microbial decomposition
12 (Goulden et al. 1998; Dutta et al. 2006; Schuur et al. 2008; Tarnocai et al. 2009; Vonk et al.
13 2013; Mann et al. 2014). Thaw depths of the seasonally-thawed active layer rapidly respond to
14 warming air temperatures (Hinkel and Nelson 2003; Frauenfeld et al. 2004) and active layer
15 thicknesses are projected to increase as a result of climate warming, releasing fractions of
16 previously frozen OC from near-surface permafrost (Zhang et al. 2005; Frey and McClelland
17 2009).

18 Deepening active layers are projected to increase the degree of interaction between soils
19 and water (Neff et al. 2006; Battin et al. 2008; Davydov et al. 2008; Frey and McClelland 2009),
20 allowing water-soluble fractions of recently-thawed OC to dissolve into soil water. The
21 permafrost underlying the active layer prevents the subsurface flow from percolating deeper
22 (Wickland et al. 2007), facilitating soil water export which, in turn, may lead to an increase in
23 the export of permafrost OC to inland waters (Vonk et al. 2013; Spencer et al. 2015). Dissolved

1 fractions of terrestrial OC can fuel microbial respiration and the production of greenhouse gasses
2 (Ågren et al. 2008; Battin 2008; Wang et al. 2014). When terrestrial dissolved organic carbon
3 (DOC) enters aquatic systems, a portion is further mineralized to CO₂, which can escape to the
4 atmosphere or be assimilated by plants and algae living in the water (Tao and Lin 2000; Vonk et
5 al. 2013). Other potential fates of terrestrially-derived DOC in aquatic systems are export via
6 transportation, sequestration via flocculation and sedimentation, or bioassimilation by microbes
7 and organisms through aquatic food webs (Cole et al. 2007; McGuire et al. 2010).

8 Up to 40% (210-456 Pg C; Strauss et al. 2013; Walter Anthony et al. 2014) of permafrost
9 soil OC is stored in ice-rich loess-dominated soils referred to as *yedoma*. Formed in unglaciated
10 regions of Siberia, Alaska, and NW Canada during the late Pleistocene (Soloviev 1959; Zimov et
11 al. 2006a; Kanevskiy et al. 2011), yedoma soils are thick (<50 m) silt-dominated deposits rich (2-
12 30%) in OC for mineral soils (Schirrmeister et al. 2011a). Yedoma is extensive in NE Siberia,
13 where it underlies an area of over 1,000,000 km² and averages 25 m in thickness (Romanovskii
14 1993; Zimov et al. 2006a). Evidence shows yedoma deposits are currently thawing (Romanovsky
15 et al. 2010), but the potential for greenhouse gas production by microbial decomposition of
16 organic matter in thawed yedoma soils has been studied with limited geographic and spatial
17 scope (Zimov et al. 1997; Zimov et al. 1993; Dutta et al. 2006; Walter et al. 2007a; Lee et al.
18 2012; Mann et al. 2012; Knoblauch et al. 2013; Vonk et al. 2013).

19 Ancient (20,000-35,800 yr bp) DOC from yedoma soils has found to be more biolabile
20 than Arctic stream, river, and permafrost DOC from non-yedoma systems (Vonk et al. 2013;
21 Mann et al. 2014; Drake et al. 2015; Spencer et al. 2015). However, it has been suggested that
22 the OC content of yedoma soils is not evenly distributed across geographic space and
23 stratigraphic layers (Dutta et al. 2006; Zimov et al. 2006a and 2006b; Schirrmeister et al. 2011a).

1 Respiration rates of ancient (21,700 yr bp) DOC from a single yedoma outcrop in the Kolyma
2 River Basin, NE Siberia were found to be 1.3-1.6 times higher than those of modern DOC
3 collected from different sites in the same yedoma-dominated watershed (Mann et al. 2014). This
4 suggests that location and particular permafrost soil forming processes could affect the
5 magnitude of potential C release from thawing permafrost. In order to estimate the potential
6 magnitude of gas production from thawing yedoma, the soil OC content of which is
7 geographically variable, it is important to gain a better understanding the potential respiration
8 rates of water-soluble soil OC collected from a wide distribution of sites.

9 Water-soluble OC derived from soil extracts is representative of the most mobile and
10 labile fractions of soil OC (Ohno et al. 2009; He et al. 2011) which can be readily utilized by
11 microbes (Matzner and Borken 2008; Wang et al. 2014). Therefore, in permafrost-dominated
12 regions, examining C respiration rates from soil extracts collected from a variety of
13 environments with varying amounts of disturbance and available OC provides a useful
14 perspective as to how climate change may alter C cycling dynamics in these systems. The
15 objective of this study was to quantify the variability of water-soluble soil OC respiration
16 potentials from 16 active layer and thawed shallow (<1.5 m) permafrost soils from five different
17 landscapes in the Kolyma River Basin, NE Siberia, Russia using laboratory incubations of soil
18 extracts mixed with river water as a proxy for water-soluble permafrost OC biolability in inland
19 waters. This study was performed as part of the National Science Foundation's POLARIS Project
20 (www.thepolarisproject.org), a summer research program for undergraduate students. Despite a
21 small sample size and limited geographic scope, we hypothesized that the C respiration
22 potentials of incubated extracts would be directly related to soil OC content and that the OC

1 contents and C respiration potentials would be spatially variable on scales of both stratigraphic
2 layers and geographic locations.

3 **Materials and methods**

4 *Study site and sample collection.* The Kolyma River Basin is the largest arctic river basin
5 entirely underlain by continuous permafrost, spanning ~650,000 km² across NE Siberia (Vtyurin
6 1975; Griffin et al. 2011; Holmes et al. 2011). The Kolyma River Basin is largely underlain by
7 yedoma permafrost and the DOC in its rivers shifts from modern ($\Delta^{14}\text{C} > 100\%$) C in the spring
8 to older ($\Delta^{14}\text{C} < 0\%$) C in the fall, suggesting the origin of the DOC transitions from shallow to
9 deeper soils as the active layer seasonally deepens (Neff et al. 2006). We collected soil samples
10 at five sites in the Kolyma River Basin in July 2010 (Table 1; Figure 1). The sites were selected
11 to be representative of various yedoma-dominated landscapes in the mixed forest region of the
12 Kolyma River watershed. Duvyanni Yar is a ~30-40 m tall yedoma outcrop exposed by the
13 Kolyma River through thermokarst and thermoerosion. Shuchi Lake and Tube Dispenser Lake
14 are first generation thermokarst (thaw) lakes formed in thick yedoma permafrost (>40 m) within
15 5 km of the town of Cherskii. The Rodinka soil pit (160 m above sea level) was dug in the Finish
16 Creek valley on the SW flank of Rodinka mountain, where the yedoma horizon is relatively thin
17 (~15 m), near faded thermokarst and solifluction features. The Bulldozer site refers to a lower
18 elevation slope beneath Rodinka mountain, 60 m above sea level. The Bulldozer site is a field of
19 residual thermokarst mounds (baydzherakhs) where the surface four meters of the soil profile
20 have thawed following bulldozer excavation and removal of the surface organic horizon and
21 active layer (60-80 cm) in 2003. The surface at our study soil profiles at Duvyanni Yar and
22 Shuchi Lake had not been disturbed previously. In contrast, bulldozer and natural thermokarst
23 activity had removed much of the overlying peat prior to our sampling at the Tube Dispenser

1 Lake, Rodinka, and Bulldozer study sites. We report surface vegetation species at each site in the
2 Supplementary Information (Table S1). With the exception of the Duvyanni Yar site, none of the
3 sample sites exhibited signs of cryoturbation (Figure S1 in Supplementary Information).

4 We delineated soil profiles at each sample location from either cross-sectional permafrost
5 exposures or soil pits (Table 1). We determined the thickness of the active layer at these sites by
6 probing to the depth of permafrost at the time of sampling (July), which was prior to the
7 maximum thickness of seasonally thawed active layer (September). Samples to examine soil
8 characteristics (gravimetric soil moisture, bulk density, and organic matter content) were
9 collected in ~10 cm intervals at each soil profile. Samples for soil extract chemistry and
10 respiration experiments were collected from four depths along soil profiles representative of
11 different potential regions for soil-water interactions. The top (10-15 cm) and bottom (30-50 cm)
12 of the seasonally thawed active layer, determined using July thaw depths, represent regions of
13 near-surface soils which presently experience seasonal freeze-thaw cycles that process and
14 degrade OC. The transient layer (70-100 cm), consisting of permafrost soils thawed
15 approximately 7,000-5,000 y.a. during the Holocene thermal optimum that subsequently re-froze
16 under colder climate conditions (Sher et al. 1979; Schirrmeister et al. 2011b), represents near-
17 surface soils which do not thaw seasonally today but may seasonally thaw in a future warmer
18 climate. Yedoma permafrost, sampled at depths greater than 100 cm below the ground surface
19 and representing older Pleistocene-aged permafrost that has not thawed during the Holocene
20 based on the presence of massive ice wedges adjacent to our sampling. These samples represent
21 near-surface permafrost soils where the OC has not been previously degraded by freeze-thaw
22 cycles during the Holocene. The depth of the boundary between the transient layer and yedoma
23 permafrost was determined by assuming the tops of visible ice wedges represented the maximum

1 historic active layer thaw depth at that site. In total we collected 16 (4 top active layer, 4 bottom
2 active layer, 4 transient layer, and 4 yedoma; Table 2) samples for soil extract chemistry and
3 respiration experiments. Each of these samples was divided into two subsamples and stored
4 (holding time ≤ 10 days) in the dark at 15 °C prior to analysis.

5 *Soil characteristics.* We analysed all of the soil characteristics samples ($n = 44$) and one
6 of each soil extract chemistry and respiration experiment subsample pair ($n = 16$) for gravimetric
7 soil moisture, bulk density, and organic matter content. Gravimetric water content was
8 determined as the difference in mass between the soil at field moisture and the oven-dried (105
9 °C for 48 hours) soil over the mass of the oven-dried soil. Bulk density was measured as the dry
10 soil mass divided by the soil subsample volume. We determined soil organic matter content by
11 calculating the mass difference between the oven-dried and ashed (400 °C for 4 hours) soil. In
12 our calculations we assumed soil OC content was fifty percent of soil organic matter mass
13 (Pribyl 2010). Grain size distribution was determined using standard hydrometer methods
14 (Interstate Standard 2008), with results presented in the Supplementary Information (Table S2).

15 *Soil extract chemistry.* Soil extracts were prepared from the second of each respiration
16 experiment soil subsample pair ($n = 16$) by vigorously mixing a 100 g subsample of soil at field
17 moisture with 1 L deionized water for 30 minutes. The extract was filtered through a
18 precombusted (450 °C for 4 hours) glass microfiber filter (0.7 μm , Whatman GF/F) to remove
19 particulate organic matter and analyzed for ammonium ($\text{NH}_4\text{-N}$; detection limit 5 $\mu\text{g L}^{-1}$) using a
20 fluorometric method (Taylor et al. 2007) and soluble reactive phosphorous ($\text{PO}_4\text{-P}$; detection
21 limit 10 $\mu\text{g L}^{-1}$) using the molybdenum-blue method (Rigler 1966). Dissolved organic carbon
22 (DOC) content was quantified using a Shimadzu TOC-V using established protocols (Mann et al.

1 2012). We calculated the fraction of water soluble OC as the ratio between the mass of DOC
2 extracted and the mass of total OC in the bulk soil utilized to make the extract.

3 *Respiration experiments.* We conducted heterotrophic respiration experiments on
4 permafrost soil extracts mixed with water from the Panteleikha River as a proxy for water-
5 soluble OC bioavailability in inland waters. The Panteleikha River drains an area of 1,500 km²
6 and is a tributary to the Kolyma River. Panteleikha River water was obtained from ~1 m depth
7 on July 27, 2010. We analysed a subsample of river water for DOC, NH₄-N, and PO₄-P using the
8 methods described in the previous section. Respiration potentials were measured by incubating
9 60 mL of the prepared soil extract with 240 mL of unfiltered Panteleikha River water within
10 standard 300 mL biological oxygen demand (BOD) bottles (Wheaton). Duplicate experimental
11 incubation vials were prepared for each of the 16 soil extract samples. Two BOD bottles
12 containing 300 mL unfiltered Panteleikha River water were prepared as an experimental control.
13 All incubation vials were tightly sealed and maintained in the dark at 20 °C, the approximate
14 temperature of the Panteleikha River's near-surface water in July.

15 We calculated biological oxygen (O₂) demand using standard methods as the loss of
16 dissolved O₂ over a five-day period (APHA 1992). We determined dissolved O₂ concentrations
17 in the bottles using a benchtop BOD O₂ probe (YSI 556; accuracy ± 0.2 mg L⁻¹); none of the
18 bottles reached anoxia within the five-day period. Oxygen consumption was converted to C
19 respiration by assuming that all dissolved O₂ loss was due to aerobic respiration and each unit
20 loss in O₂ corresponded to a unit production of carbon dioxide (CO₂). The calculated CO₂
21 production over the incubation period was multiplied by the ratio of carbon's atomic mass to
22 oxygen's atomic mass (0.375) to determine the mass of C respired in each incubation vial.

1 We report the C respiration of each sample in terms of: total C respired (mg), net C
2 respired from the soil extract (mg), the mass of C respired per gram soil OC (mg C g OC^{-1}), and
3 the fraction of total DOC respired (%). Total C respiration from the BOD bottle is reported as
4 cumulative C respired from both the river water and the soil extracts during the five-day BOD
5 incubation period. Net C respiration from the soil extract is reported as the total C respiration
6 from the BOD bottle minus the mean C respiration measured in the river water controls. The
7 mass of C respired per gram soil OC was calculated by dividing the net C respiration from the
8 soil extract by the mass of OC in the soil used to prepare the extract. We calculated the fraction
9 of DOC respired by dividing the total C respiration by the measured amount of total DOC within
10 the BOD bottle. All respiration data are presented as mean \pm standard error (SE) together with a
11 reported sample size, where n represents the number of soil extract samples. An n value of 1
12 represents the mean of duplicate laboratory incubation vials containing extract from a single field
13 soil sample.

14 *Statistics.* All statistical analyses were conducted using MATLAB (R2013a Student
15 Version) software. Soil OC contents and all soil extract data ($\text{NH}_4\text{-N}$, $\text{PO}_4\text{-P}$, DOC, net C
16 respiration, C respiration g soil OC^{-1} , % DOC respired) were tested for normal distribution using
17 the Jarque-Bera test. The data for DOC, net C respiration, and % DOC respired were found to be
18 consistent with a normal distribution at the $\alpha = 0.05$ level. Data for soil OC, soil extract $\text{NH}_4\text{-N}$
19 and $\text{PO}_4\text{-P}$ concentrations, and C respiration g soil OC^{-1} were not consistent with a normal
20 distribution at the $\alpha = 0.05$ level; these variables were log-transformed to improve normality.
21 Since half of our data parameters were found to be inconsistent with a normal distribution, we
22 determined the statistical significance of differences in soil OC and extract parameters between
23 soil layers and sampling sites using the nonparametric Mann-Whitney U-test. We determined all

1 correlations using Spearman's rank correlation coefficients. Both the differences and correlations
2 were considered statistically significant when $p \leq 0.05$ ($\alpha = 0.05$ confidence level). Finally, we
3 conducted forward stepwise multiple linear regressions to examine how well our tested soil
4 [log(OC)] and extract [DOC, log(NH₄-N), log(PO₄-P)] parameters predicted the measured C
5 respiration [net C respiration, log(C respiration g soil OC⁻¹), and % DOC respired]. Terms were
6 added to the stepwise regression models using the standard SSE criterion.

7 **Results**

8 *Soil characteristics.* Soil dry bulk density ranged from 0.21-1.13 g cm⁻³ (median 0.70 g cm⁻³,
9 mean 0.67 ± 0.23 g cm⁻³, n = 44) and gravimetric water content ranged from 7.2-70.7% moisture
10 (median 24.0%, mean $28.1 \pm 16.7\%$, n = 45; Figure S2). Soil bulk OC content ranged from 1-
11 20.5% by mass (median 1.7%, mean $2.7 \pm 3.3\%$ OC, n = 46). Soils from Duvyanni Yar (1.7-
12 20.3% OC, median 2.9%, mean $4.9 \pm 5.5\%$, n = 11) and Shuchi Lake Ridge (1.2-12.9% OC,
13 median 1.2%, mean $2.7 \pm 4.1\%$, n = 8), the two sites which had not experienced prior
14 disturbance, had higher levels of bulk soil OC ($p = 0.021$ and 0.001 , respectively) compared to
15 other sampling sites. There were no additional statistically significant differences in soil
16 characteristics between sampling sites or soil stratigraphic layers.

17 *River water and soil extract chemistry.* River water from the Panteleikha River had initial
18 nutrient concentrations of 50.0 ± 5 µg L⁻¹ NH₄-N, 310.0 ± 10 µg L⁻¹ PO₄-P, and 4.83 mg L⁻¹
19 DOC (Table 2). Ammonium (NH₄-N) concentrations extracted from the soil samples ranged
20 from 24.8-4,573 µg L⁻¹ NH₄-N (median 250 µg L⁻¹, mean 747 ± 437 µg L⁻¹, n = 11). Extracted
21 soluble reactive phosphorous (PO₄-P) concentrations ranged from 7.3-90.4 µg L⁻¹ PO₄-P (median
22 28.6 µg L⁻¹, mean 31.7 ± 6.4 µg L⁻¹, n = 12). Dissolved organic carbon (DOC) concentrations in
23 the soil extract varied by an order of magnitude, ranging from 2.8 to 27.9 mg L⁻¹ (median 9.1 mg

1 L⁻¹, mean 13.5 ± 2.5 mg L⁻¹, n = 14). Soils with higher OC contents extracted more DOC (r =
2 0.58, p = 0.029). Fractions of water soluble OC in our extracts ranged from 0.15-1.78% (median
3 0.57%, mean 0.71 ± 0.14% OC, n = 14). There were no statistically significant differences in soil
4 extract chemistry between sampling sites or soil stratigraphic layers. Full river water and soil
5 extract chemistry results are shown in Table 2.

6 *Carbon respiration.* Total C respiration during the five-day incubation period ranged
7 from 0.17-0.67 mg C (median 0.28 mg, mean 0.33 ± 0.08 mg, n = 16; Table 3). The fraction of
8 total DOC (river water DOC plus soil extract DOC) respired during the incubation period ranged
9 from 8.9-31.4% (median 17.8%, mean 18.2 ± 1.7%, n = 14). Soil extract respiration ranged from
10 -0.03 to 0.47 mg C (median 0.08 mg, mean 0.13 ± 0.08 mg, n = 16) and -0.42 to 5.21 mg C g soil
11 OC⁻¹ (median 0.62 mg C g soil OC⁻¹, mean 1.32 ± 1.77 mg C g soil OC⁻¹, n = 16). Extract
12 samples from the Shuchi Lake Ridge bottom active layer, Duvyanni Yar bottom active layer, and
13 Tube Dispenser Lake transitional layer respired less C than the river water control. Overall, the
14 bottom active layer had lower C respiration potentials (total C respired and C respired from soil
15 extract) than the other stratigraphic layers (p = 0.012 for both; Figure 2). There were no
16 statistically significant differences in C respiration between sampling sites (Figure S3). Full C
17 respiration results are shown in Table 3.

18 Correlation analyses did not show any statistically significant relationships between C
19 respiration, soil OC contents, and extract DOC, NH₄-N, or PO₄-P concentrations. Forward
20 stepwise multiple linear regression analyses revealed that the mass of soil extract C respired (mg)
21 and the fraction of total DOC respired (%) could be estimated by the log-transformed NH₄-N
22 concentrations in the soil extracts (p = 0.046 and 0.043, respectively; Table 4). Carbon

1 respiration in terms of C respired per g soil OC was not predicted by any measured parameters in
2 the final linear model. Complete data for the linear regression models are presented in Table 4.

3 **Discussion**

4 Our results indicate that soils in the Kolyma River Basin, NE Siberia are spatially variable in
5 terms of soil OC content. Soils from Duvyanni Yar and Shuchi Lake Ridge had significantly ($p <$
6 0.05) higher OC contents compared to the other sampled sites. We hypothesize this is due to lack
7 of prior disturbance removing much of the surface vegetation, allowing modern plants and roots
8 to provide greater fresh OC input at these two sites. The observed variance in yedoma
9 (Pleistocene permafrost) OC content across sites (1.5–3.0% OC) suggests that soil OC is
10 unevenly distributed across the yedoma-permafrost dominated landscape. This outcome would
11 be expected given that, even within a single site, soil OC content can vary by an order of
12 magnitude throughout the vertical profile of tens of meters due to paleoenvironmental
13 differences during the Pleistocene and Holocene (Tarnocai et al. 2009; Schirrmeister 2011a).

14 Samples collected from the bottom active layer had lower mean C respiration rates than
15 the other stratigraphic layers in our study ($p = 0.012$). Soils in the bottom active layer currently
16 experience seasonal freeze-thaw cycles, which degrades soil OC and promotes increased water-
17 soluble OC release (Wang et al. 2014). Although soils in the top active layer also experience
18 annual freezing and thawing, the top active layer receives inputs of fresh, labile OC from modern
19 plants which are densely rooted in this layer, while soils in the bottom active layer presumably
20 receive lower inputs of fresh organic matter from modern plants. In addition, deeper active layer
21 depths in late summer cause increased water residence times in the bottom active layer compared
22 to the top active layer, which increases soil-water interaction and can potentially facilitate OC
23 dissolution and export (Neff et al. 2006; Wickland et al. 2007; Frey and McClelland 2009).

1 However, it is important to note that in natural systems this process is highly watershed-
2 dependent and, depending on local environmental and permafrost conditions, increased soil-
3 water interaction can also lead to the sorption of DOC to mineral soils (McDowell and Wood
4 1984; Neff and Asner 2001; Frey and McClelland 2009).

5 The C respiration rates of shallow yedoma (Pleistocene silt-dominated permafrost) in our
6 study were highly variable by site. The yedoma soil extract from Duvyanni Yar respired 3.4
7 times more C than the averaged yedoma extracts from the Tube Dispenser Lake, Rodinka, and
8 Bulldozer sites, although its soil OC content was on average only 1.3 times higher. This suggests
9 that the yedoma OC was more bioavailable at the Duvyanni Yar site and yedoma organic matter
10 quality is not homogenous across the Kolyma River Basin, which concurs with the findings of
11 Mann et al. (2014) concerning the spatial variability of DOC biolability in the same watershed.
12 We did not observe statistically significant differences in C respiration potentials when
13 comparing the transient layer and the yedoma. We had hypothesized that extracts from the
14 transient layer would respire less C than extracts from the Pleistocene permafrost due to the most
15 labile fractions of its OC pool being previously degraded by freeze-thaw cycles and microbial
16 respiration during the Holocene thermal optimum (Sher et al. 1979; Schirrmeister et al. 2011b;
17 Vonk et al. 2013; Wang et al. 2014). While the lack of a statistically significant difference in C
18 respiration rates may be due to our small sample size, it is also possible that near-surface yedoma
19 with the potential to thaw from deepening active layer thicknesses may not have significantly
20 higher proportions of bioavailable OC when compared to the overlying shallow transient layer
21 permafrost.

22 We observed lower respiration rates in 25% of vials containing soil extracts compared to
23 the control vials containing only river water (Table 3), suggesting that the soil extracts in these

1 samples may contain inhibitory organic compounds that reduce DOC decomposition in aquatic
2 systems. For instance, phenolic compounds have been found to inhibit both bacterial abundance
3 and microbial metabolism, even in the presence of high OC and nutrients (Fenner and Freeman
4 2011; Mann et al. 2014). While we did not measure OC composition or quality in our study,
5 these results suggest soil OC composition must also be taken into account when estimating
6 potential C respiration potentials from recently-thawed permafrost.

7 Previous studies have shown that permafrost disturbance, including thermokarst activity,
8 can export large amounts of labile DOC to inland waters (Vonk et al. 2013; Abbott et al. 2014).
9 However, in our study there were no statistically significant differences in C respiration between
10 previously undisturbed profiles (Duvyanni Yar and Shuchi Lake Ridge) and sites where the
11 surface had been previously disturbed (Bulldozer Site, Tube Dispenser Lake, Rodinka). This is
12 consistent with findings from Larouche et al. (2015), which showed that labile DOC fractions in
13 Alaskan inland waters were not significantly altered by disturbance. Previously undisturbed sites
14 in our study had higher OC contents in their profiles, and higher OC contents in bulk soil
15 samples positively correlated with higher DOC levels in our soil extracts ($r = 0.58$, $p = 0.029$).
16 Although the fractions of water-soluble OC in our soil samples were low (0.15-1.78%), the
17 proportions of soil OC extracted in our study were consistent with general proportions of water-
18 extractable soil organic matter in total soil organic matter (2-5%; Ellerbrock et al. 1999; He et al.
19 2011). While higher soil OC contents correlated with higher DOC concentrations in our soil
20 extracts, higher extract DOC concentrations did not lead to higher C respiration potentials and
21 soil extract C respiration potentials did not correlate with soil OC contents in our study. Our
22 finding that higher extract DOC concentrations did not necessarily yield higher C respiration

1 potentials contradicts previous studies in which higher DOC concentrations in inland waters led
2 to increased release of CO₂ (Algesten et al. 2005; Battin et al. 2008; Lapierre et al. 2013).

3 One possible explanation is that C respiration in our study is limited by an alternative
4 factor other than DOC concentrations. For instance, macronutrient availability, organic matter
5 composition, and microbial communities within our incubations may have influenced extract C
6 respiration rates observed in this study. Prior studies have found that landscape-scale variation in
7 soil C bioavailability can be linked to soil C to N ratios, with higher values leading to more C
8 respiration (Schuur et al. 2015). In aquatic systems, nutrient concentrations and ratios in the
9 environment could limit C respiration when they deviate from the Redfield ratio (16 N : 1 P) or
10 microbial N:P biomass ratios (7 N: 1 P; Cleveland and Liptzin 2007). Previous studies in arctic
11 aquatic ecosystems found bacterial production is usually limited by phosphorous (Granéli et al.
12 2004; Hobbie and Laybourn-Parry 2008), but we did not find a statistically significant positive
13 relationship between extract C respiration potentials and PO₄-P concentrations in our study. Our
14 multiple linear regression analyses indicated positive correlation between C respiration and
15 ammonium (NH₄-N) concentrations in the soil extracts, suggesting potential N limitation in our
16 system. However, since ammonium is only a fraction of the total N pool, we are unable to
17 determine if there is significant correlation between C respiration and C to N ratios in our study.

18 Approximately 8.9-31.2% of the total DOC in our incubation vials was respired during
19 our 5-day incubation period. This is consistent with several prior laboratory incubations which
20 found that only 4-60% of DOC from soils is available for microbial respiration (Kalbitz et al.
21 2003; Holmes et al. 2008; Vonk et al. 2013; Larouche et al. 2015; Spencer et al. 2015). While we
22 acknowledge the limitations of using laboratory incubations to extrapolate what proportion of
23 water-soluble soil OC will be respired upon entering inland waters, we note that our incubation

1 period ($t = 5$ days) is consistent with mean residence times it takes water to move from
2 permafrost thaw sites to the Kolyma River main stem (3-7 days; Vonk et al. 2013). In the
3 Kolyma River Basin and other inland water sites, the proportion of soil OC available for
4 microbial respiration, and by extension the soil's net C respiration potential, increases as initial
5 processing and breakdown by ultraviolet light makes OC compounds more bioavailable for
6 heterotrophic respiration and the production of CO_2 (Cory et al. 2013). Other studies suggest up
7 to 22% of DOC can be sequestered by reactive iron and thus made unavailable for C respiration
8 (Salvadó et al. 2015). Therefore, further research is necessary to understand the full C respiration
9 potential of water-soluble OC from the Kolyma River yedoma region's thawing permafrost as it
10 is processed *in situ*. *Conclusions*. Examining soil OC quantity and quality can lead to a better
11 understanding of how much soil C can be immediately processed following permafrost thaw and
12 water-soluble OC release to inland waters. Our study indicates that, while soil OC content is
13 spatially variable in the Kolyma River Basin, it is not necessarily an indicator of C respiration
14 potentials upon permafrost thaw and export to inland waters. Sites which had not experienced
15 prior disturbance had higher soil OC contents, but there were no statistically significant
16 differences in C respiration potentials between disturbed v. undisturbed sites in our study. This
17 suggests alternative OC quality parameters, including exposure to freeze-thaw cycling, the extent
18 of seasonal interactions with water, and the presence of inhibitory organic compounds, may also
19 play a role in water-soluble C respiration potentials. It is also important to consider factors which
20 may alter *in situ* C respiration from thawing permafrost, such as DOC sorption to mineral soils
21 and the additional physical procession DOC experiences after being released to inland waters,
22 when estimating the true magnitude of C release from thawing permafrost soils.

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9 **References**

- 10 Abbot, B.W., et al. 2014. Elevated dissolved organic carbon biodegradability from thawing and
11 collapsing permafrost. *Journal of Geophysical Research: Biogeosciences*, JG002678,
12 doi:10.1002/2014JG002678.
- 13 Ågren, A., et al. 2008. Terrestrial export of highly bioavailable carbon from small boreal
14 catchments in spring floods. *Freshwater Biology*, 53, 964-72, doi:10.1111/j.1365-
15 2427.2008.01955.x.
- 16 Algesten, G., et al. 2005. Contribution of Sediment Respiration to Summer CO₂ Emission from
17 Low Productive Boreal and Subarctic Lakes. *Microbial Ecology*, 50(4), 529-535,
18 doi:10.1007/s00248-005-5007-x.
- 19 APHA, 1992. Standard methods for the examination of water and wastewater. 18 ed.
20 Washington, DC.
- 21 Battin, T. J. K., et al. 2008. Biophysical controls on organic carbon fluxes in fluvial networks.
22 *Nature Geosci*, 1(2), 95-100, doi:10.1038/ngeo101.
- 23 Brown, J., et al. 1998, Revised 2001. Circum-Arctic map of permafrost and ground ice
24 conditions. Boulder, CO: National Snow and Ice Data Center/World Data Center for
25 Glaciology.
- 26 Cleveland, C., and Liptzin, D., 2007. C:N:P stoichiometry in soil: is there a “Redfield ratio” for
27 the microbial biomass? *Biogeochemistry*, 85(3), 235-252, doi:10.1007/s10533-007-9132-
28 0.
- 29 Cole, J., et al. 2007. Plumbing the Global Carbon Cycle: Integrating Inland Waters into the
30 Terrestrial Carbon Budget. *Ecosystems*, 10(1), 172-185, doi:10.1007/s10021-006-9013-8.

- 1 Cory, R. M., et al. 2013. Surface exposure to sunlight stimulates CO₂ release from permafrost
2 soil carbon in the Arctic. *Proceedings of the National Academy of Sciences of the United*
3 *States of America*, 110(9), 3429-3434, doi:10.1073/pnas.1214104110.
- 4 Cory, R. M., et al. 2014. Sunlight controls water column processing of carbon in arctic fresh
5 waters. *Science*, 345(6199), 925-928, doi:10.1126/science.1253119.
- 6 Davydov, S., et al. 2008. Changes in Active-Layer Thickness and Seasonal Fluxes of Dissolved
7 Organic Carbon as a Possible Baseline for Permafrost Monitoring. *Proceedings of the IX*
8 *Conference on Permafrost*. University of Alaska Fairbanks, USA June 29–July. 3, 2008,
9 333-337.
- 10 Drake T.W., et al. 2015. Ancient low-molecular-weight organic acids in permafrost fuel rapid
11 carbon dioxide production upon thaw. *Proceedings of the National Academy of Sciences*,
12 112(45), 13946–13951, doi: 10.1073/pnas.1511705112.
- 13 Dutta, K., et al. 2006. Potential carbon release from permafrost soils of Northeastern Siberia.
14 *Global Change Biology*, 12(12), 2336-2351, doi:10.1111/j.1365-2486.2006.01259.x.
- 15 Ellerbrock, R. H., et al. 1999. Characterization of soil organic matter from a sandy soil in relation
16 to management practice using FTIR spectroscopy. *Plant and Soil*, 213, 55-61.
- 17 Fenner, N., and Freeman, C., 2011. Drought-induced carbon loss in peatlands. *Nature Geosci*,
18 4(12), 895-900, doi:10.1038/ngeo1323.
- 19 Frauenfeld, O.W., et al. 2004. Interdecadal changes in seasonal freeze and thaw depths in Russia.
20 *Journal of Geophysical Research*, 109, D05101, doi: 10.1029/2003JD004245.
- 21 Frey, K.E., and McClelland, J.W., 2009. Impacts of permafrost degradation on arctic river
22 biogeochemistry. *Hydrological Processes*, 23, 169-182, doi:10.1002/hyp.7196.
- 23 Goulden, M. L., et al. 1998. Sensitivity of boreal forest carbon balance to soil thaw. *Science*,
24 279(5348), 214-217, doi:10.1126/science.279.5348.214.
- 25 Granéli, W., et al. 2004. Phosphorus limitation of bacterial growth in high Arctic lakes and
26 ponds. *Aquatic Sciences- Research Across Boundaries*, 66(4), 430-439,
27 doi:10.1007/s00027-004-0732-7.
- 28 Griffin, C. G., et al. 2011. Spatial and interannual variability of dissolved organic matter in the
29 Kolyma River, East Siberia, observed using satellite imagery. *Journal of Geophysical*
30 *Research*, 116, G03018, doi:10.1029/2010JG001634.

- 1 He, Z., et al. 2011. Elemental and fourier thransform-infared spectroscopic analysis of water- and
2 pyrophosphate-extracted soil organic matter. *Soil Science*, 176(4), 183-189,
3 doi:10.1097/SS.0b013e318212865c.
- 4 Hinkel, K.M., and Nelson, F.E., 2003. Spatial and temporal patterns of active layer thickness at
5 Circumpolar Active Layer Monitoring (CALM) sites in northern Alaska, 1995–2000.
6 *Journal of Geophysical Research- Atmospheres*, 108(D2), 8168,
7 doi:10.1029/2001JD000927.
- 8 Hobbie, J. E., and Laybourn-Parry, J., 2008. Heterotrophic microbial processes in polar lakes.
9 *Polar Lakes and Rivers*. Oxford, Great Britain: Oxford University Press, 197-212.
- 10 Holmes, R.M., et al. 2008. Lability of DOC transported by Alaskan rivers to the Arctic Ocean.
11 *Geophysical Research Letters*, 35, L03402, doi:10.1029/2007GL032837.
- 12 Holmes, R. M., et al. 2011. Seasonal and Annual Fluxes of Nutrients and Organic Matter from
13 Large Rivers to the Arctic Ocean and Surrounding Seas. *Estuaries and Coasts*, 35(2),
14 369-382, doi:10.1007/s12237-011-9386-6.
- 15 Hugelius, G., et al. 2014. Estimated stocks of circumpolar permafrost carbon with quantified
16 uncertainty ranges and identified data gaps. *Biogeosciences*, 11(23), 6573-6593,
17 doi:10.5194/bg-11-6573-2014.
- 18 Interstate Standard, 2008. GOST 12536-79. Soils. Methods of laboratory granulometric (grain-
19 size) and microaggregate distribution. Moscow. IPK Izdatelstvo Standartov. 18 p.
- 20 IPCC, 2013. Working Group I Contribution to the IPCC Fifth Assessment Report. *Climate*
21 *Change 2013: The Physical Science Basis*.
- 22 Kalbitz, K., et al. 2003. Biodegradation of soil-derived dissolved organic matter as related to its
23 properties. *Geoderma*, 113(3-4), 273–291, doi:10.1016/S0016-7061(02)00365-8.
- 24 Kanevskiy, M., et al. 2011. Cryostratigraphy of late Pleistocene syngenetic permafrost (yedoma)
25 in northern Alaska, Ikillik River exposure. *Quaternary Research*, 75(3), 584-596,
26 doi:10.1016/j.yqres.2010.12.003.
- 27 Knoblauch, C., et al. 2013. Predicting long-term carbon mineralization and trace gas production
28 from thawing permafrost of Northeast Siberia. *Global Change Biology*, 19, 1160-1172,
29 doi:10.1111/gcb.12116.

- 1 Lapierre, J.-F., et al. 2013. Increases in terrestrially-derived carbon stimulate organic carbon
2 processing and CO₂ emissions in Canadian aquatic ecosystems. *Nature Communications*,
3 4, Article 2972, doi:10.1038/ncomms3972.
- 4 Larouche, J.R., et al. 2015. The role of watershed characteristics, permafrost thaw, and wildfire
5 on dissolved organic carbon biodegradability and water chemistry in Arctic headwater
6 streams. *Biogeosciences*, 12, 4221-4233, doi: 10.5194/bg-12-4221-2015.
- 7 Lee, H., et al. 2012. The rate of permafrost carbon release under aerobic and anaerobic
8 conditions and its potential effects on climate. *Global Change Biology*, 18, 515-527,
9 doi:10.1111/j.1365-2486.2011.02519.x.
- 10 Mann, P.J., et al. 2012. Controls on the composition and lability of dissolved organic matter in
11 Siberia's Kolyma River basin. *Journal of Geophysical Research*, 117, G01028,
12 doi:10.1029/2011JG001798.
- 13 Mann, P.J., et al. 2014. Evidence for key enzymatic controls on metabolism of Arctic river
14 organic matter. *Global Change Biology*, 20, 1089-1100, doi:10.1111/gcb.12416.
- 15 Matzner, E., and Borken, W., 2008. Do freeze–thaw events enhance C and N losses from soils of
16 different ecosystems? A review. *Eur. J. Soil. Sci.*, 59, 274–284, doi:10.1111/j.1365-
17 2389.2007.00992.x.
- 18 McDowell, W.H., and Wood, T., 1984. Podzolization: soil processes control dissolved organic
19 carbon concentrations in stream water. *Soil Science*, 137, 23-32.
- 20 McGuire, A. D., et al. 2010. The carbon budget of the northern cryosphere region. *Current*
21 *Opinion in Environmental Sustainability*, 2(4), 231-236,
22 doi:10.1016/j.cosust.2010.05.003.
- 23 Neff, J. C., et al. 2006. Seasonal changes in the age and structure of dissolved organic carbon in
24 Siberian rivers and streams. *Geophysical Research Letters*, 33, L23401,
25 doi:10.1029/2006GL028222.
- 26 Neff, J.C., and Asner, G.P., 2001. Dissolved organic carbon in terrestrial ecosystems: synthesis
27 and a model. *Ecosystems*, 4, 29-48, doi:10.1007/s100210000058.
- 28 Ohno, T., et al. 2009. Influence of tillage, cropping, and nitrogen source on the chemical
29 characteristics of humic acid, fulvic acid, and water-soluble soil organic matter fractions
30 of a longterm cropping system study. *Soil Science*, 174(12), 652-660,
31 doi:10.1097/SS.0b013e3181c30808.

- 1 Pribyl, D. W., 2010. A critical review of the conventional SOC to SOM conversion factor.
2 Geoderma, 156(3-4), 75-83, doi:10.1016/j.geoderma.2010.02.003.
- 3 Rigler, F. H., 1966. Radiobiological analysis of inorganic phosphorus in lakewater. Verh. Int.
4 Ver. Theor. Angew. Limnol., 16, 465-470.
- 5 Romanovskii, N. N., 1993. Osnovy kriogeneza litosfery (Basics of cryogenesis of lithosphere).
6 Moscow: Izdatelstvo, Moscow State University, 296-313.
- 7 Romanovsky, V., et al. 2010. Permafrost thermal state in the polar Northern Hemisphere during the
8 International Polar Year 2007-2009: a synthesis. Permafrost and Periglacial Processes, 21(2),
9 106-116, doi:10.1002/ppp.689.
- 10 Schirrmeister, L., et al. 2011a. Fossil organic matter characteristics in permafrost deposits of the
11 northeast Siberian Arctic. Journal of Geophysical Research, 116, G00M02,
12 doi:10.1029/2011JG001647.
- 13 Schirrmeister, L., et al. 2011b. Sedimentary characteristics and origin of the Late Pleistocene Ice
14 Complex on north-east Siberian Arctic coastal lowlands and islands – A review.
15 Quaternary International, 241(1/2), 3-25, doi:10.1016/j.quaint.2010.04.004.
- 16 Schuur, E. A. G., et al. 2008. Vulnerability of permafrost carbon to climate change: Implications
17 for the global carbon cycle. Bioscience, 58(8), 701-714, doi:10.1641/B580807.
- 18 Schuur, E.A.G., et al. 2015. Climate change and the permafrost carbon feedback. Nature, 520,
19 171-179, doi:10.1038/nature14338.
- 20 Sher, A.V., et al. 1979. Late Cenozoic of the Kolyma Lowland: XIV Pacific Science Congress,
21 Tour Guide XI, Khabarovsk August 1979. Moscow: Academy of Sciences of the USSR.
- 22 Strauss, J., et al. 2013. The deep permafrost carbon pool of the yedoma region in Siberia and
23 Alaska. Geophysical Research Letters, 40(23), 6165-6170,
24 doi:10.1002/2013GL058088.
- 25 Striegl, R. G., et al. 2005. A decrease in discharge-
26 normalized DOC export by the Yukon River during summer through autumn.
27 Geophysical Research Letters, 32(21), L21413, doi:10.1029/2005GL024413.
- 28 Soloviev, P. A., 1959. The cryolithozone in the north part of the Lena-Amga-Interfluve.
29 Academy of Science of the USSR.
- 30 Spencer, R.G.M., et al. 2015. Detecting the signature of permafrost thaw in Arctic rivers.
Geophysical Research Letters, 42, doi:10.1002/2015GL063498.

- 1 Tao, S., and Lin, B., 2000. Water soluble organic carbon and its measurement in soil and
2 sediment. *Water Research*, 34(5), 1751-1755, doi:10.1016/S0043-1354(99)00324-3.
- 3 Tarnocai, C., et al. 2009. Soil organic carbon pools in the northern circumpolar permafrost
4 region. *Global Biogeochemical Cycles*, 23, GB2023, doi:10.1029/2008GB003327.
- 5 Vtyurin, B.I., 1975. *Ground ice of the USSR*. Nauka, Moscow (Russian).
- 6 Vonk, J. E., et al. 2013. High biolability of ancient permafrost carbon upon thaw. *Geophysical*
7 *Research Letters*, 40(11), 2689–2693, doi:10.1002/grl.50348.
- 8 Walter Anthony, K.M., et al. 2014. A shift of thermokarst lakes from carbon sources to sinks
9 during the Holocene epoch. *Nature*, 511, 452-456, doi:10.1038/nature13560. Walter, K.
10 M., et al. 2007a. Thermokarst Lakes as a Source of Atmospheric CH₄ During the Last
11 Deglaciation. *Science*, 318(5850), 633-636, doi:10.1126/science.1142924.
- 12 Walter, K. M., et al. 2007b. Methane bubbling from northern lakes: present and future
13 contributions to the global methane budget. *Philosophical Transactions of the Royal*
14 *Society a-Mathematical Physical and Engineering Sciences*, 365(1856), 1657-1676,
15 doi:10.1098/rsta.2007.2036.
- 16 Wang, J., et al. 2014. CO₂ emissions from soils of different depths of a permafrost peatland,
17 Northeast China: response to simulated freezing–thawing cycles. *Journal of Plant*
18 *Nutrition and Soil Sciences*, 177(4), 524-531, doi:10.1002/jpln.201300309.
- 19 Wickland, K.P., et al. 2007. Dissolved organic carbon in Alaskan boreal forest: sources,
20 chemical characteristics, and biodegradability. *Ecosystems*, (10) 1323-1340,
21 doi:10.1007/s10021-007-9101-4.
- 22 Zhang, T., et al. 2005. Spatial and temporal variability in active layer thickness over the Russian
23 Arctic drainage basin. *J. Geophys. Res.*, 110, D16101, doi:10.1029/2004JD005642.
- 24 Zimov, S. A., et al. 1993. Wintertime CO₂ Emission from Soils of Northeastern Siberia. *Arctic*,
25 46(3), 197-204.
- 26 Zimov, S. A., et al. 2006a. Permafrost and the global carbon budget. *Science*, 312(5780), 1612-
27 1613, doi:10.1126/science.1128908.
- 28 Zimov, S.A., et al. 2006b. Permafrost carbon: Stock and decomposability of globally significant
29 carbon pool. *Geophysical Research Letters*, 33, L20502, doi:10.1029/2006GL027484.
- 30 Zimov, S. A., et al. 1997. North Siberian Lakes: A methane source fueled by Pleistocene carbon.
31 *Science*, 277(5327), 800-802, doi:10.1126/science.277.5327.800.

1 **Tables**

2 Table 1. Location, sampling dates, profile delineation method, total depth of profile, depth to the
 3 permafrost table, and prior observable disturbance at the time of sampling for the soil profiles in
 4 this study. See Table S1 in the Supplementary Information for surface vegetation cover at each
 5 site.

Site	Lat.	Long.	Date sampled	Profile delineation method	Profile depth (cm)	Depth to permafrost table (cm)	Prior surface disturbance
Shuchi Lake Ridge	68.748°N	161.384°E	8 July 2010	Soil pit	93	55	None
Duyanni Yar	68.630°N	159.154°E	20 July 2010	Thermokarst exposure	370	35	None
Bulldozer Site	68.698°N	161.539°E	9 July 2010	Thermokarst exposure	80	- ^a	Removal of surface 60-80 cm by bulldozer activity in 2003
Tube Dispenser Lake	68.897°N	161.407°E	12 July 2010	Soil pit	190	70	Thermokarst activity
Rodinka	68.724°N	161.588°E	14 July 2010	Thermokarst exposure	190	103	Thermokarst activity and thermal erosion

^a Did not reach permafrost table in this profile

6

1 Table 2. Soil sample depths, bulk density, field moisture content, organic carbon (OC) content, water extractable OC fractions, and measured
 2 dissolved organic carbon (DOC), ammonium (NH₄-N), and orthophosphate (PO₄-P) concentrations for all soil extracts and Panteleikha River
 3 water from different sampling sites and stratigraphic layers in the Kolyma River Basin. Each line in the table represents one soil sample collected
 4 and processed as described in the Materials and methods.

Sample		Soil Properties					Soil Extract Chemistry		
Site	Stratigraphic layer ^a	Depth (cm)	Dry bulk density (g cm ⁻³)	Field moisture (%)	OC content (%)	Fraction water extractable OC (%)	DOC (mg L ⁻¹)	NH ₄ -N (µg L ⁻¹)	PO ₄ -P (µg L ⁻¹)
Shuchi Lake Ridge	TAL	15	0.78	35	13.0	0.33	27.85	- ^b	- ^b
	BAL	50	0.60	18	1.5	0.45	5.55	24.83	33.79
	TL	90	0.45	19	1.5	0.23	2.82	2608	36.67
Duvyanni Yar	TAL	15	0.97	65	20.5	0.27	19.15	118.0	7.29
	BAL	35	0.49	22	2.5	0.15	3.01	255.1	59.30
	TL	70	0.22	48	4.0	1.06	22.11	360.0	23.47
Bulldozer Site Tube	PP	150	0.56	36	2.5	1.02	16.37	4573	10.26
	PP	80	0.82	10	1.5	0.68	9.13	74.11	47.49
Dispenser Lake	TAL	10	0.74	24	1.0	1.19	9.03	87.30	BDL ^c
	BAL	30	0.74	24	1.0	- ^b	- ^b	- ^b	- ^b
	TL	70	0.74	47	1.5	- ^b	- ^b	- ^b	28.61
Rodinka	PP	110	0.35	36	1.5	0.84	8.10	249.8	90.37
	TAL	15	0.75	12	2.5	0.23	5.08	- ^b	- ^b
	BAL	40	0.67	20	2.5	0.36	7.28	81.95	22.88
Panteleikha River	TL	100	0.70	38	2.5	1.78	27.66	- ^b	28.61
	PP	120	1.04	38	3.0	1.40	25.95	533.4	22.57
Panteleikha River	River Water	~100	N/A	N/A	N/A	N/A	4.83	50.00	310.0

^a TAL = Top Active Layer; BAL = Bottom Active Layer; TL = Transient Layer; PP = Pleistocene Permafrost

^b Parameter was not measured.

1 Table 3. Mass of initial dissolved organic carbon (DOC) in the BOD bottles, the initial DOC partitioning between soil extract DOC and river
 2 water DOC, the amount of carbon (C) respired during the five day incubation \pm standard error between duplicate bottles, C respired from the soil
 3 extracts (net C respiration – river water control), C respired per g soil OC (C respired from soil extract / mass soil OC used to make the extract),
 4 the fraction of total DOC (river DOC + extract DOC) respired, and the effect of the soil extract on C respiration. The effect of the soil extract on
 5 C respiration was determined as positive (+) or negative (-) based on if C respiration from the soil extracts was higher or lower than the river
 6 water control, respectively.

Sample		DOC			C Respiration				
Site	Stratigraphic layer ^a	DOC contribution from soil extract (mg C)	DOC contribution from river water (mg C)	Net initial DOC in BOD bottle (mg C)	Total C respired in BOD bottle (mg C)	C respired from soil extract (mg C)	C respired per g soil OC (mg C g OC ⁻¹)	Fraction of total DOC respired (%)	Effect of soil extract on C respiration ^c
Panteleikha River	River Water	N/A	1.45	1.45	0.20 \pm 0.01	N/A	N/A	13.82	N/A
Shuchi Lake Ridge	TAL	1.67	1.16	2.83	0.50 \pm 0.05	0.30	0.59	17.48	+
	BAL	0.33	1.16	1.49	0.17 \pm 0.02	-0.03	-0.41	11.34	-
	TL	0.17	1.16	1.33	0.35 \pm 0.03	0.15	2.06	26.02	+
Duvyanni Yar	TAL	1.15	1.16	2.31	0.44 \pm 0.02	0.24	0.56	19.17	+
	BAL	0.18	1.16	1.34	0.19 \pm 0.00	-0.01	-0.09	14.05	-
	TL	1.33	1.16	2.49	0.45 \pm 0.03	0.25	2.00	18.12	+
	PP	0.98	1.16	2.14	0.67 \pm 0.27	0.47	4.90	31.35	+
Bulldozer Site	PP	0.55	1.16	1.71	0.24 \pm 0.01	0.04	0.50	14.33	+
Tube Dispenser Lake	TAL	0.54	1.16	1.70	0.36 \pm 0.01	0.16	3.51	20.99	+
	BAL	- ^b	1.16	- ^b	0.24 \pm 0.01	0.04	0.88	- ^b	+
	TL	- ^b	1.16	- ^b	0.18 \pm 0.01	-0.02	-0.42	- ^b	-
Rodinka	PP	0.49	1.16	1.65	0.50 \pm 0.02	0.30	5.21	30.28	+
	TAL	0.30	1.16	1.46	0.30 \pm 0.09	0.10	0.76	20.62	+
	BAL	0.44	1.16	1.60	0.20 \pm 0.01	0.00	0.00	12.82	--

TL	1.66	1.16	2.82	0.26 ±0.13	0.06	0.66	9.28	+
PP	1.56	1.16	2.72	0.24 ±0.02	0.04	0.36	8.86	+

^aTAL = Top Active Layer; BAL = Bottom Active Layer; TL = Transient Layer; PP = Pleistocene Permafrost

^bParameter was not measured.

^c+ indicates increased respiration; - indicates suppressed respiration; -- indicates no change

1 Table 4. Results from forward stepwise multiple linear regression analyses examining C
 2 respiration as the response variable in terms of soil extract C respired (mg), C respired per
 3 gram soil OC (mg C g OC⁻¹), and the fraction total DOC respired (%). Soil OC contents and
 4 soil extract DOC and log(PO₄-P) concentrations did not predict C respiration in any model.
 5 Soil extract log(NH₄-N) concentrations predicted C respiration in terms of soil extract C
 6 respired (mg) and the fraction total DOC respired (%). Carbon respiration per gram soil OC
 7 had no significant predictor variables in the final regression model.

Parameter		C Respiration		
		Soil extract C respired (mg C)	C respired per g soil OC (mg C g OC ⁻¹)	Fraction total DOC respired (%)
β ₀ [Intercept]	Value	-0.760	1.149	0.502
	SE	0.545	0.085	7.811
	p-value	0.201	0.113	0.950
β ₁ [log(NH ₄ -N)]	Value	0.221	- ^a	3.225
	p-value	0.046	- ^a	0.043
	SE	0.094	- ^a	1.341
Regression Model	df	8	9	8
	RMSE	0.45	0.27	6.46
	R ²	0.411	- ^b	0.420
	p	0.046	- ^b	0.043

^a Parameter was not in final regression model
^b No parameters were significant (p ≤ 0.05) in final regression model

8

1 **Figure Captions**

2 Figure 1. Map showing the five soil sample sites in the Kolyma River Basin, Northeast
3 Siberia, Russia (a) and photographs showing locations of the sample sites, marked using
4 white triangles (b-f). Surface vegetation data for each site are presented in Table S1 in the
5 Supplementary Information. The Bulldozer site (b), located in the Rodinka mountain
6 piedmont, is in a field of residual thermokarst mounds (baydzherakhs) with the Kolyma and
7 Panteleikha Rivers floodplain visible in the distance. The Tube Dispenser Lake profile (c)
8 was collected near exposures of melting ice wedges on the southern slope of the Tube
9 Dispenser Lake, a thermokarst lake. The Rodinka site (d), located on the lower slope of
10 Rodinka mountain, was collected near faded thermokarst and solifluction forms in the Finish
11 Creek valley. The flourishing violet, yellow, and white flowers indicate the site is in the first
12 succession stadium following completion of active thermokarst and thermal erosion. The
13 Shuchi Lake Ridge site (e) is located within a “drunken forest,” formed due to yedoma ice
14 complex degradation and sediment sliding into the southern wall of the thermokarst Shuchi
15 Lake. Finally, the Duvyanni Yar site (f) was collected from the Duvyanni Yar thermokarst
16 and thermoerosional exposure, cut by the Kolyma River, of the Late Pleistocene ice complex.
17 Photographs (b-f) by V.V. Spektor.

18 Figure 2. Carbon respiration separated by site (left) and by stratigraphic layer (right). The top
19 graphs show the mass of C respired from the soil extracts (mg C); the middle graphs show the
20 mass of C respired per g soil OC (mg C g OC⁻¹); the bottom graphs show the fraction of total
21 DOC respired during the 5-day incubation period (%).